

**SECTION OF ANALYTICAL CHEMISTRY
COMMISSION ON MICROCHEMICAL TECHNIQUES***

**RECOMMENDED TEST SUBSTANCES FOR THE
MICRODETERMINATION OF CARBON
AND HYDROGEN**

A number of compounds are recommended for use as test substances for the microdetermination of carbon and hydrogen in organic substances. All of these substances, or a proper selection from the list, may be used to determine the universal applicability of a given method, either already described or one which might be developed in the future.

The compounds selected are stable over long periods of time and are non-hygroscopic (any exceptions to the latter generalization are noted). The substances are either commercially available in a sufficiently pure state to be used for tests based on the accuracy of present-day methods, or may be purified or prepared by conventional laboratory means to meet these standards.

The compounds selected include the following:

- (1) those which have extreme values (high as well as low) for carbon and/or hydrogen contents;
- (2) compounds representing a variety of structural types;
- (3) compounds containing elements which may cause interferences in the carbon–hydrogen determination.

These recommendations may be changed or supplemented as the need arises.

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C _n H _m	Empirical formula	Molecular weight	C (%)	H (%)	O (%)	N (%)	S (%)	Element (%)
C ₁₀ H ₁₆	C ₄ H ₁₀ C ₅ H ₁₂ C ₁₀ H ₈	178.234 84.162 128.174	94.34 85.63 93.71	5.66 14.37 6.29				
O (C _n H _m)	C ₂₇ H ₄₆ O ₂ C ₂₇ H ₄₆ O C ₈ H ₁₆ O ₆ C ₈ H ₁₆ O ₃ C ₁₈ H ₃₆ O ₂	122.125 386.665 180.162 148.120 284.486	68.85 83.87 40.00 64.87 76.00	4.95 11.99 6.71 2.72 12.76	26.20 4.14 53.59 32.41 11.25			
N (C _n H _m O)	C ₈ H ₈ ON C ₈ H ₁₀ O ₂ N ₄	135.168 194.200	71.09 49.48	6.71 5.19	11.84 16.48	10.36 28.85		
Acetanilide								
Caffeine (1,3,7-trimethyl-xanthine anhydrous)	C ₈ H ₁₀ N ₄	198.146	36.37	3.05	32.30	28.28		
2,4-Dinitrophenylhydrazine	C ₁₂ H ₁₄ N ₄	169.228	85.17	6.55		8.28		
Diphenylamine	C ₁₂ H ₁₂ N ₄	140.194	51.40	8.63		39.97		
Hexamethylenetetramine	C ₆ H ₁₂ N ₄	229.114	31.45	1.32	48.88			
Picric acid	C ₆ H ₃ O ₇ N ₃							
F (C _n H _m O _z)	C ₂ H ₂ O ₂ F C ₄ F ₈ C ₈ H ₆ ONF ₃	140.117 662.154 189.144	60.00 25.39 50.80	3.60	22.84			
Fluorobenzoic acid								
Perfluorodicyclohexylethane								
Trifluoroacetonilide					8.46	7.41		
Cl (C _n H _m)	C ₂ H ₃ ONCl C ₅ H ₁₀ N ₂ Cl C ₅ Cl ₆ C ₆ H ₆ Cl ₆ C ₆ O ₂ Cl ₄	93.519 202.563 284.808 290.856 245.894	25.69 35.58 25.30 24.78 29.31	4.31 1.49	17.11 31.60	14.98 13.83	Cl (%)	
Chloroacetamide							37.91	
1-Chloro-2,4-dinitrobenzene							17.50	
Hexachlorobenzene							74.70	
Hexachlorocyclohexane							75.14	
Tetrachloro- <i>p</i> -benzoquinone (chloranil, tetrachloroquinone)							57.68	
Br (C _n H _m)	C ₆ H ₅ ONBr C ₆ H ₃ OBr ₃	214.076 330.838	44.88 21.78	3.77 0.91	7.47 4.84	6.54	Br (%)	
<i>p</i> -Bromacetanilide							37.33	
2,4,6-Tribromophenol							72.47	

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I (C ₁₀ H ₈ ,..)	C ₂₀ H ₈ O ₅ I ₄	835.924	28.74 <u><u>2</u></u>	0.96	9.57	I (%) 60.3
Erythrosin (iodoquinophorescin) 2,4,5,7-tetraiodofluorescein)	C ₇ H ₈ O ₂ I	248.027	33.90 <u><u>2</u></u>	2.03	12.90	Cl (%) 51.17
s-Iodoacetic acid						17.49
S (C ₆ H ₅ ,..)	C ₈ H ₁₁ N ₂ ClS	202.715	47.40	5.47	13.82	15.82
s-Benzylthiourea chloride	C ₆ H ₈ O ₂ N ₂ S	172.212	41.85	4.68	16.27	18.62
Sulphanilamide	C ₆ H ₇ O ₃ NS	173.196	41.61	4.07	27.71	18.51
Sulphanilic acid (anhydrous)	C ₇ H ₁₆ O ₂ S ₂	228.337	36.82	7.06	28.03	28.09
Sulphonamides	CH ₄ N ₂ S	76.125	15.78	5.30	36.80	42.12
Thiourea						
P, As (C ₆ H ₅ ,..)	C ₆ H ₈ O ₃ NAs	217.048	33.20	3.72	22.11	6.45
o-Arsanilic acid (o-aminophenylarsonic acid)	C ₉ H ₁₄ O ₆ N ₂ ClSP	344.725	31.36 <u><u>2</u></u>	4.09	27.85	9.30
5-Chloro-4-hydroxy-3-methoxybenzyl isothiourea phosphate*	C ₁₈ H ₁₅ P	262.293	82.43 <u><u>2</u></u>	5.76	8.13	10.29
Triphenylphosphine						
Me (C ₆ H ₅ ,..)	C ₂ O ₄ Ca	128.102	18.75		49.96	11.81
Calcium oxalate						
Phenylmercuric acetate	C ₈ H ₈ O ₂ Hg	336.762	28.53	2.39	9.50	Ca (%) 31.29
Potassium acid phthalate	C ₈ H ₅ O ₄ K	204.228	47.05	2.47	31.34	Hg (%) 59.57
Sodium oxalate	C ₂ O ₄ Na ₂	134.004	17.93		47.76	K (%) 19.15
						Na (%) 34.31

* W. H. Smith. *Anal. Chem.*, **30**, 149 (1958).

Notes

For the calculation of molecular weights and percentages, the atomic weights used are those proposed by the Commission on Atomic Weights of the I.U.P.A.C., 1957.

The percentage figures are given to the second decimal. Where the third decimal is less than 0.005, the last figure has been disregarded. Where the last figure has been rounded off, it has been underlined (0.375 = 0.38).